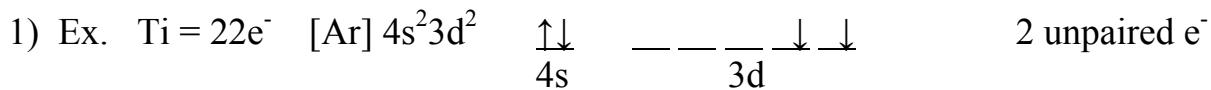


Name \_\_\_\_\_

## Unit 2-7: Orbital Notation

For each, write the abbreviated electron configurations, then draw the orbital notation of the non-noble gas portion of the electron configuration, and then tell how many unpaired electrons are present.



2) Mg

3) Cl

4) Zn

5) Se

6) Ba

7) Nb

8) Os

9) Pb

10) Fr